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sandviches! _

Introduction: When we
bake/cook something,

we use a specific
amount of each

ingredient. Imagine if
you made a batch of

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cookies and used way too many eggs, or not enough sugar.

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LabLet's make some sandwiches!

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When we bake/cook something, we use a specific amount of each

ingredient. Imagine if you made a batch of cookies and used way too many eggs, or not enough sugar. In chemistry, reactions proceed with very specific recipes.

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The study of these recipes is stoichiometry. When the reactants are present in ...

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PhET Lab - SobTell

Answer to Basic Stoichiometry Post-Lab Homework Exercises Load the 'Reactants, Products, and Leftovers' simulation and work through each of the levels of the...

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Limiting Reactants

Intro - PhET

Contribution

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through an intuitive, game-like environment where students learn through exploration and discovery.

PhET Interactive Simulations - PhET: Free online physics

...

Create your own sandwich and then see how many sandwiches you can make with different amounts of ingredients. Do the same with chemical

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reactions. See how many products you can make with different amounts of reactants. Play a game to test your understanding of reactants, products and leftovers. Can you get a perfect score on each level?

Reactants, Products and Leftovers - PhET

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we use a specific

amount of each

ingredient.

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Example |

Graduateway

forming the question,

or need help seeing

how the lab relates to

stoichiometry;

performing the

stoichiometry; special

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care should be spent making sure students are using the acetic acid mass, not the mass of the vinegar. To save time I have made this Stoichiometry lab answer key so I can quickly check student work. creating a step-by-step procedure

Eleventh grade

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Chemistry
& Chemical Reactivity
6th Ed. Kotz, Treichel
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Brookes-Cole, 2006 /
ISBN:

978-0-534-99766-3

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Answer Key

Reactants, Products
and Leftovers

**Reactants, Products
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**Balancing Chemical
Equations Phet Lab**

Answer Key

Reactants, Products
and Leftovers

**Reactants, Products
and Leftovers**

The Alkaline Earth
Metals consist of six
elements which are
Beryllium, Magnesium
Calcium, Strontium,
Barium, and Radium.
The most reactive is
Radium. They are also
reactive and

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electropositive. They form basic oxides that react with water to form comparatively insoluble hydroxides. This is located on column 2.

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fully and concisely in

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the space provided.

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