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OCI 2 bent,  $104.5^\circ$ ,  $sp^3$  O Cl Cl 58. BeF 2 linear,  $180^\circ$ ,  $sp$  F Be F 59. CF 4 tetrahedral,  $109^\circ$ ,  $sp^3$  F F F F C 60.

## Covalent Bonding Covalent Bonding - Weebly

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Manganese has a body-centered cubic unit cell and has a density of  $7.88 \text{ g/cm}^3$ . From this information, determine the length of the edge of the cubic cell.

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Step-by-step solution: Step 1 of 2 Alkenes are commonly prepared via one of two routes. Step 2 of 2 The mechanisms to form each of these species are shown below: The hydroxide ion abstracts proton from carbon atom to form more substituted alkene 2-methylbut-2-ene is the major product (according to saytzeff's rule).

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22 Solutions Manual for Analytical Chemistry 2.1 2. (a) The values are as follows: mean: 243.5 mg median: 243.4 mg range:



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37.4 mg standard deviation: 11.9 mg variance: 141 (b) We are interested in the area under a normal distribution curve that lies to the right of 250 mg, as shown in Figure SM4.1.

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